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Title: Electrorefining Low Yield Possible Root Causes

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# Electrorefining Low Yield

Possible Root Causes

Frederic Ewing, PT-1 FLM

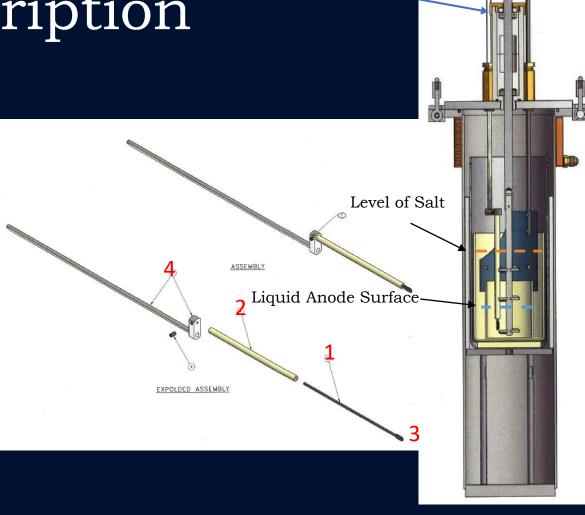
# Parts Description

(1) **Tungsten**: electrically conductive, will not oxidize significantly under applied loads (~20A, 2V), refractory metal resistant to Pu corrosion.

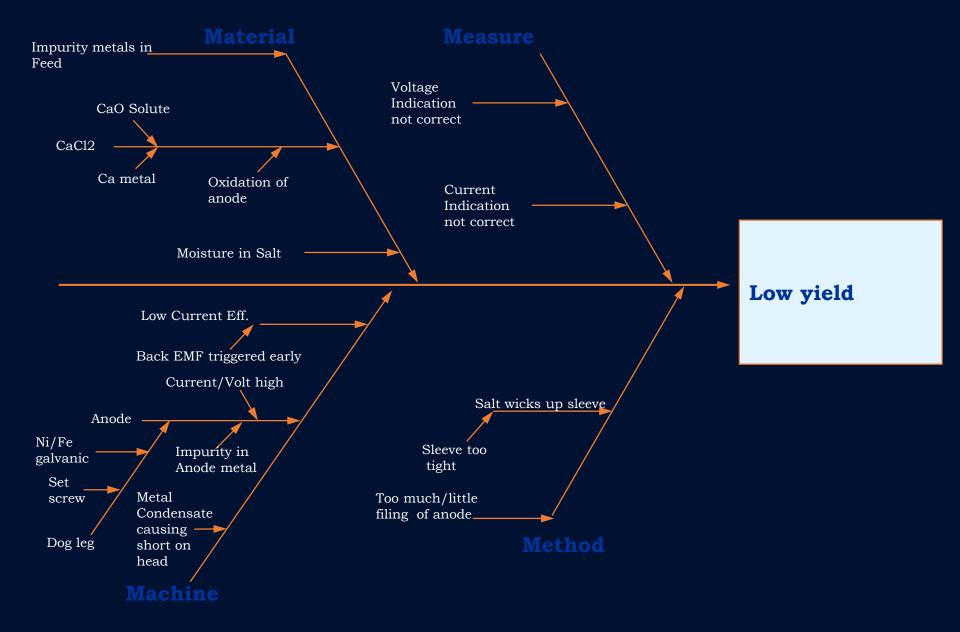
(2) **Magnesia Sleeve**: electrically insulates and protects electrode from electrolyte, same material as crucible, resistant to Pu/molten salt corrosion.

(3) **Paddle Tip**: electrical connection to liquid anode pool, prevents sleeve from falling off.

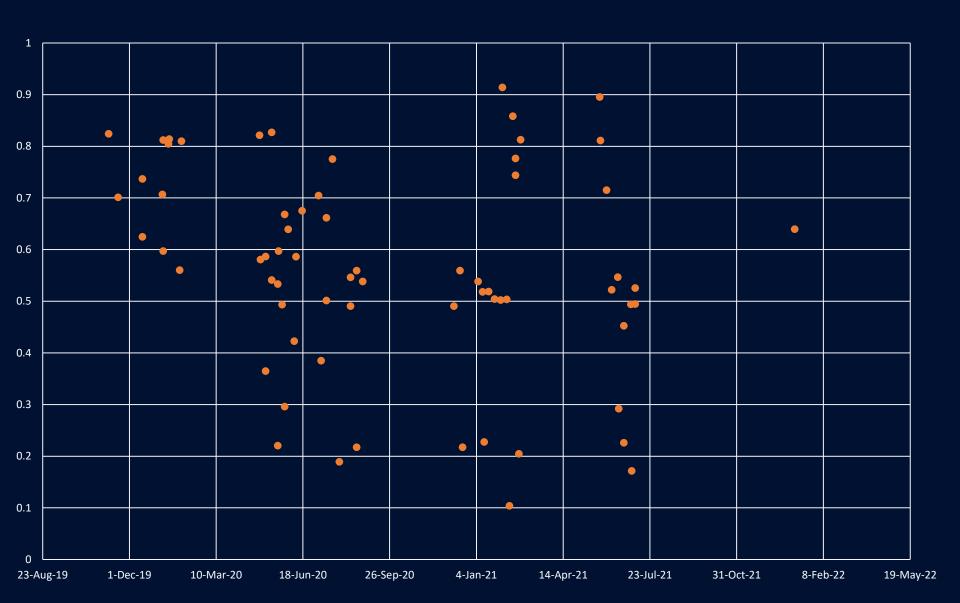
(4) **Dog Leg**: holds tungsten rod, sits inside furnace, passes current to tungsten, enables positioning rod within crucible cup.



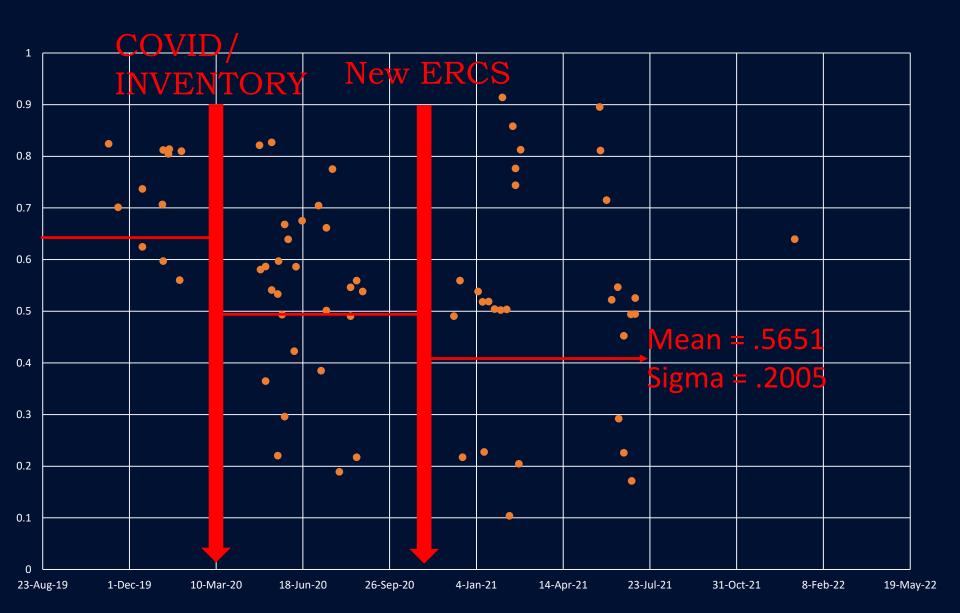
Current applied here



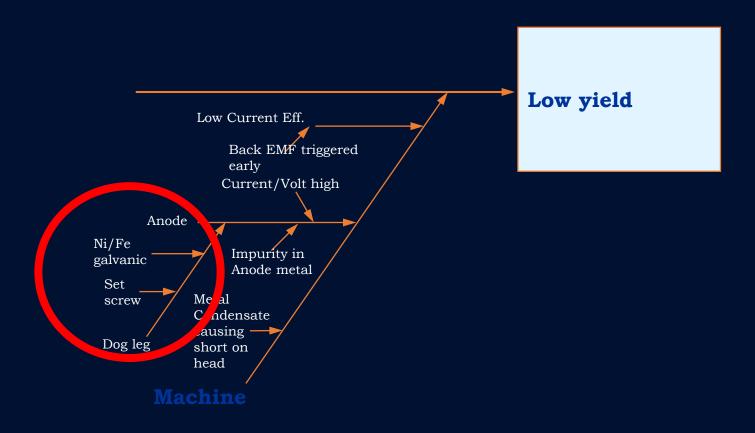
## Successful Run Yields FY 20 & 21



## Successful Run Yields FY 20 & 21







## Problem #1: Paddle Falling Off

- -Occurring in the majority of runs in the past 5 months.
- -Electrode loses contact with the liquid anode pool, passing current directly into electrolyte, prematurely ending run.
- -Sometimes, the ceramic sleeve slides off of rod when raising equipment.





# Hypothesis #1

- -Tungsten extraordinarily brittle.
- -Very difficult to fabricate with.
- -If not controlled very carefully, material properties negatively impacted (e.g., ductile to brittle transition temperature increases, grain re-crystallization)

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## OUR FABRICATION WAS NOT CONTROLLED WELL!

- -heated with acetylene torch until "red hot", leading to grain re-crystallization and brittleness.
- -press rate for paddle not controlled, resulting in crack defects.





# Hypothesis #2

-The anode was lowered into the melt too quickly!

-Thermal gradient induced stresses that fractured the sleeve.

-Thermal expansion coefficient of sleeve is higher than the tungsten. A gap was not left for the sleeve to expand, thus applying stress to sleeve when heated.

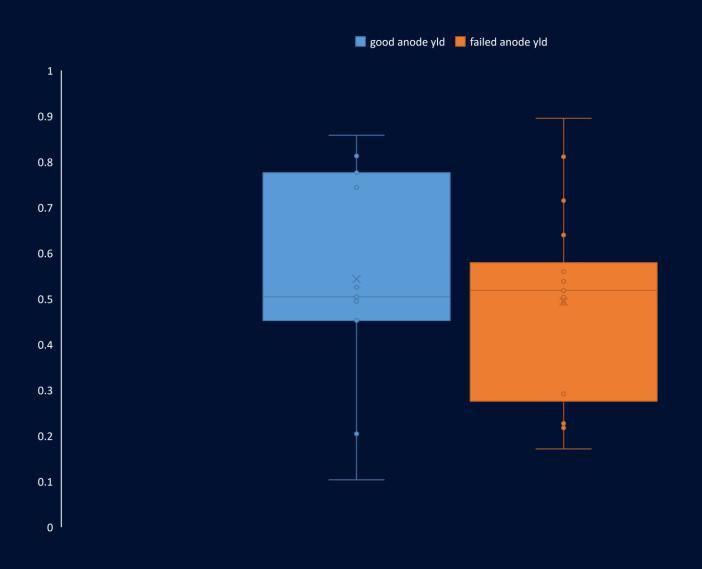
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1. Change procedure to reduce insertion rate to 0.5 inches per minute (per RFP-1876), and specify 0.125-0.25" gap when assembling anode.

2. Investigating a change from extruded magnesia, which has lower density, to different forming technique.

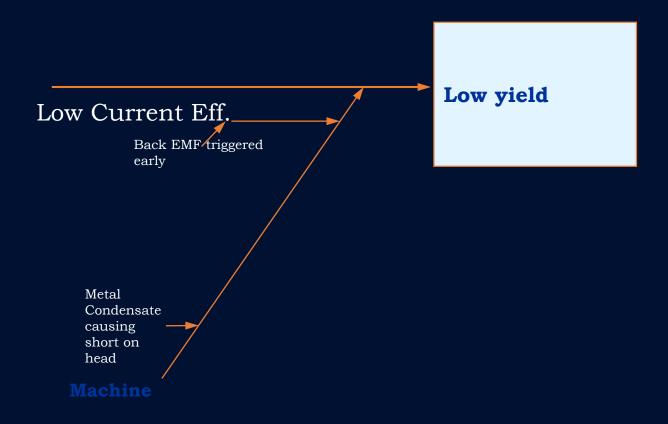


# 2021 yield failed vs good anodes



# T-Tests

t-Test: Two-Sample Assuming Unequal Variances		
	good anode yld	bad anode yld
Mean	0.543413164	0.493556767
Variance	0.058828388	0.041601097
Observations	11	18
Hypothesized Mean Difference	0	
df	18	
t Stat	0.569677385	
P(T<=t) one-tail	0.287968999	
t Critical one-tail	1.734063607	
P(T<=t) two-tail	0.575937999	
t Critical two-tail	2.10092204	
t-Test: Two-Sample Assuming Equal Variances		
	good anode yld	bad anode yld
Mean	0.543413164	0.493556767
Variance	0.058828388	0.041601097
Observations	11	18
Pooled Variance	0.047981575	
Hypothesized Mean Difference	0	
df	27	
t Stat	0.594726428	
P(T<=t) one-tail	0.27848898	
t Critical one-tail	1.703288446	
P(T<=t) two-tail	0.556977959	
t Critical two-tail	2.051830516	



## Machine

### Problem

 Metal Condensate causing short on head

Back EMF triggered early?

Filing Anode too much

#### Test

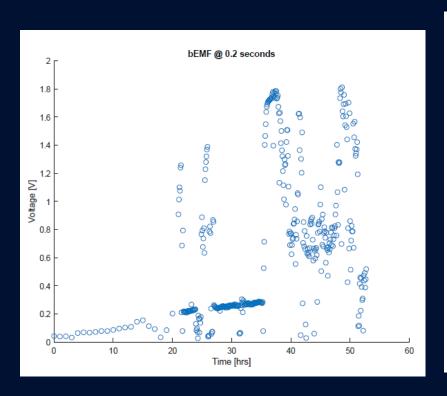
Unkown

 Raise back EMF threshold, get chemistry

• New experimental anode

# Problem #3: Premature bEMF Ramp Up

-According to ERCS data, bEMF ramps up earlier in run than historical successful run, resulting in early bEMF termination of the run.



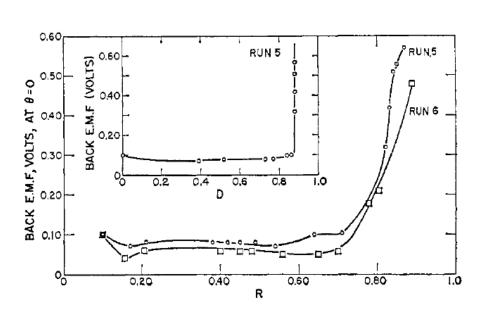
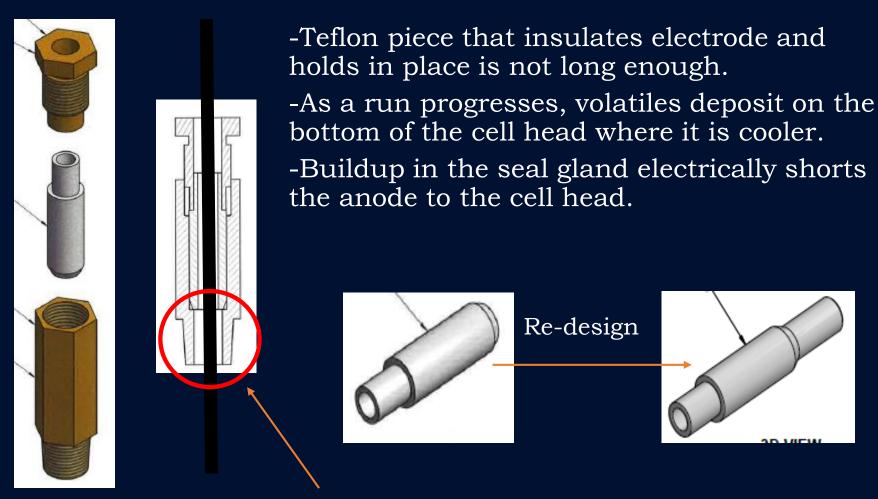


Figure 7. Back-e.m.f. at  $\theta=0$  vs. R for two typical Pu-Fe electrorefining runs

Run from this year

"Textbook" run from 1960s

# Hypothesis #3



Build-up and short location



## Measure

### Problem

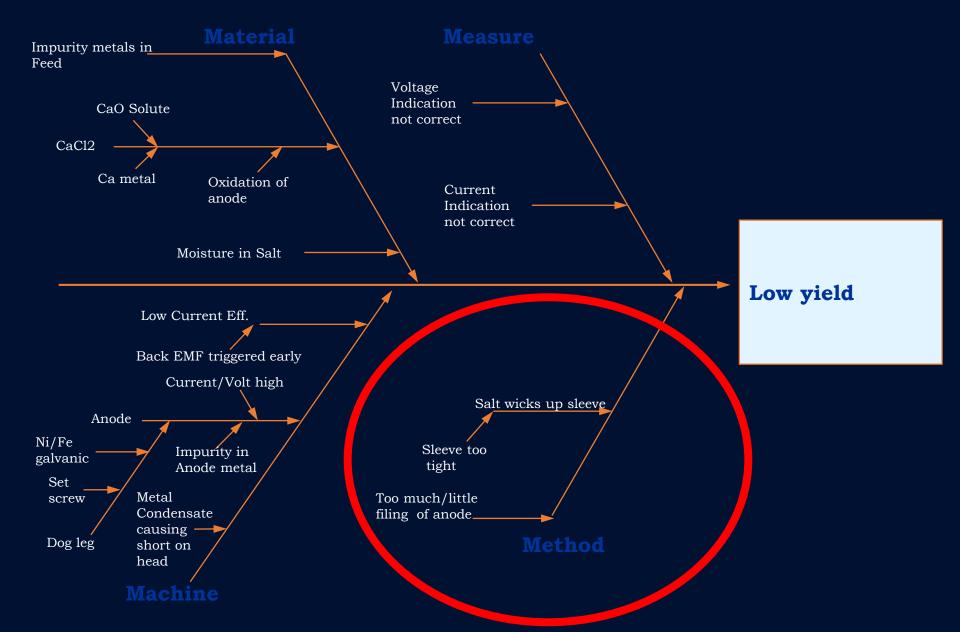
Voltage Indication incorrect

Current indication incorrect

### Test

Into multi-meter with alligator clips

 Use a current clamp on top of box



## Method

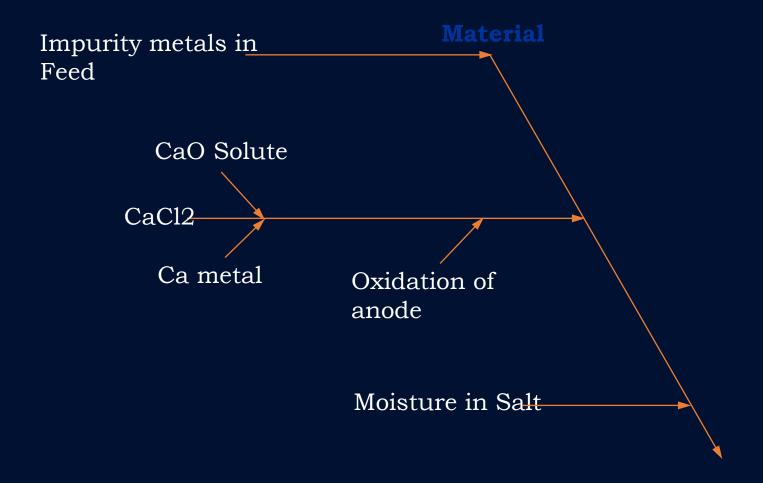
### Problem

 Sleeve is overtightened and salt wicks up sleeve and starts making contact with dog leg.

## Test

 Have a slot gauge to ensure spacing





## Material

### Problem

- All issues related to CaCl2
- Impurity in feed
- Moisture in salt

### Test

- temporarily stop using CaCl2
- Regression models chemistry data
- Use new MgCl Salt with eutectic